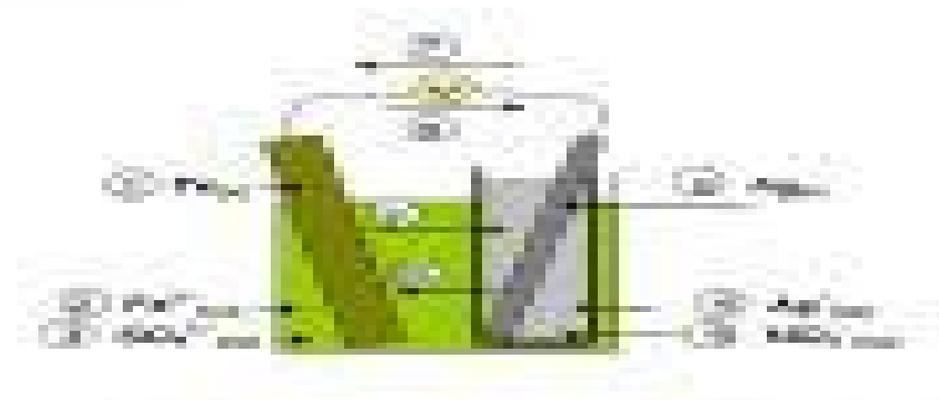


Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. metals _____
- b. metal ions _____
- c. metals _____
- d. metal ions _____
- e. electron movement flow _____
- f. displacement overall movement _____

- g. half-cell reduction equation _____
- h. Zn _____

i. Write overall cell equation with phases. $\text{cell} = \text{anode} | \text{anode solution} || \text{cathode solution} | \text{cathode}$ represent by written arguments with half-cell notation for anode & cathode

Voltaic Cells Study Guide

Yi-Tong Ma



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