

MODULE 1: ORGANIC CHEMISTRY

Organic Chemistry

- Organic compounds
- Contains carbon
 - Except CO_2 , CO_3^{2-}
- Classify according to functional group
 - Family/class
 - Similar structure and reactivity
 - Ex. CH_3OH (methanol), $\text{CH}_3\text{CH}_2\text{OH}$ (ethanol)
 - "-ol" – hydroxyl functional group

Hydrocarbons (C & H)

C-C	ALKANES	C=O	CARBONYL GROUP
C=C	ALKENES	RCOH	ALDEHYDES (CHO – Formyl)
C≡C	ALKYNES	RCOOH	CARBOXYLIC ACIDS (COOH – Carboxyl)
	AROMATIC COMPOUNDS / ARENES	CARBOXYLIC DERIVATIVES	
RX	ALKYL HALIDES	RCOOR	ESTERS
RNH ₂	AMINES	RCONH ₂	AMIDES
ROR	ETHERS	CONH ₂	CARBAMOYL
RSR	THIOETHERS (super analog of ethers)	RCOOCOR	ANHYDRIDES
ROH	ALCOHOLS	RCOX	ACYL HALIDE
RSH	THIOLS (sulfhydryl/mercapto)	RCO	ACYL
RC=OR	KETONES	CH ₃ CO	ACETYL (simplest acyl)

NOMENCLATURE

- Organic compounds may have different names
- Ex. CH_3COOH – acetic acid (semi-systematic)/ethanoic acid (systematic)/(dil.) vinegar (trivial)

IUPAC (International Uni...)

- Formulated the rules on nomenclature

Format:

- Prefix-parent-bond type-suffix
- Parent
 - Most important
 - Usually the longest chain

- Bond type
 - Single "an"
 - Double "en"
 - Triple bond "yn"
- Prefixes
 - 1 meth
 - 2 eth
 - 3 prop
 - 4 but
 - 5 pent
- Suffix & Prefix
 - Substituents
 - Suffix
 - Highest priority substituent only
 - RCOOH
 - RCOOH derivatives
 - RCHO
 - RCOR
 - ROH
 - RSH
 - RNH₂
 - C-C
 - C=C
 - C≡C
 - RX, ROR, RSR – never suffix

Ex. $\text{CH}_3\text{CH}_2\text{Cl}$

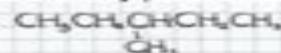
- ✓ Ethyl chloride
- ✓ Chloroethane

$\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3$

- ✓ Ethyl ether
- ✓ Ethoxyethane

Practice Set 1

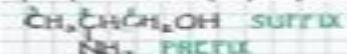
- Pentane: 3-methylpentane



- Rule: alphabetization
3-ethyl-4-methylhexane



- 2-aminopropan-1-ol (1-propanol)



- 1-chlorobut-2-en-2-ol



- for multiple subs., add multiplier prefix
1,1,2-trimethylcyclohexane



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