

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) $n = 1000$, $m = 1000$

(B) $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $1 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$
 (C) $1 \text{ mol H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$

2. An acid-base reaction involves hydrogen ions. Suppose instead you consider hydroxide ions. If 1.0 g of acid were neutralized by base, how much acid would be produced? (B, C, D are incorrect)

(A) $n = 1000000$, $m = 1000000$

(B) $1 \text{ mol H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$

3. An acid-base reaction involves H^+ ions. H^+ ions are produced by the reaction of an acid and a base. What is the pH of the solution? (A, B, C, D are incorrect)

(A) 10^{-1} , 10^{-2} , 10^{-3} , 10^{-4}

4. Which of the following is a property of the following reaction? (A, B, C, D are incorrect)



(A) 10^{-1} , 10^{-2} , 10^{-3} , 10^{-4}

5. A solution of acid and base is mixed together. The resulting solution is acidic. If 1.0 g of acid and 1.0 g of base are mixed together, what is the pH of the resulting solution? (A, B, C, D are incorrect)



(A) 10^{-1} , 10^{-2} , 10^{-3} , 10^{-4} , 10^{-5} , 10^{-6}

6. An acid-base reaction involves hydrogen ions. Suppose instead you consider hydroxide ions. If 1.0 g of acid were neutralized by base, how much acid would be produced? (A, B, C, D are incorrect)



Multiple Choice Stoichiometry Test

Brendan G. Carr



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