

# SS1 CHEMISTRY LESSON NOTES - SECOND TERM

## Scheme of Work

SN	WEEK	TOPICS
1	WEEK 1	GAS LAWS I
2	WEEK 2	GAS LAWS II
3	WEEK 3	GAS LAWS III
4	WEEK 4	STANDARD SEPARATION TECHNIQUES FOR MIXTURES I
5	WEEK 5	STANDARD SEPARATION TECHNIQUES FOR MIXTURES II
6	WEEK 6	STANDARD SEPARATION TECHNIQUES FOR MIXTURES III
7	WEEK 7	ACIDS
8	WEEK 8	BASES AND SALTS I
9	WEEK 9	BASES AND SALTS II
10	WEEK 10	WATER

# 2nd Term Chemistry Note For Ss1

**J Spring**



## **2nd Term Chemistry Note For Ss1:**

## Whispering the Techniques of Language: An Psychological Quest through **2nd Term Chemistry Note For Ss1**

In a digitally-driven world where screens reign great and quick interaction drowns out the subtleties of language, the profound techniques and emotional subtleties hidden within words often get unheard. Yet, situated within the pages of **2nd Term Chemistry Note For Ss1** a captivating fictional prize pulsing with fresh thoughts, lies an extraordinary journey waiting to be undertaken. Penned by a skilled wordsmith, that marvelous opus encourages viewers on an introspective journey, lightly unraveling the veiled truths and profound influence resonating within the fabric of each and every word. Within the emotional depths with this touching evaluation, we can embark upon a heartfelt exploration of the book is key subjects, dissect its charming writing style, and yield to the effective resonance it evokes serious within the recesses of readers hearts.

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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below normal boiling point of  $\text{NH}_3$  33.3 °C  $\Delta H_{\text{vap}}$  23.4 kJ/mol specific heat  $\text{NH}_3$  1.474 J/g K specific heat  $\text{NH}_3$  2.06 J/g K

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web apr 3 2023 the experiment described above can be summarized in a graph called a heating curve figure below figure 11 16 1 11 16 1 in the heating curve of water the temperature is shown as heat is continually added changes of state occur during plateaus because the temperature is constant

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web once all of the material has solidified at the time indicated by point c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the experimental set up by measuring cooling curves for samples of varying composition one can map the entire phase

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web for water this temperature is 100 c because the boiling point for water is 100 c different substances have different melting points and boiling points but the shapes of their heating curves are very similar for example this is the heating curve for iron a metal that melts at 1538 c and boils at 2861 c cooling curves

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