

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) $n = 1000$, $m = 1000$

(B) $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $1 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$
 (C) $1 \text{ mol H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$

2. An acid-base reaction represents a proton transfer from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The acid is stronger than the base.
 (B) The base is stronger than the acid.
 (C) The acid is stronger than the base.
 (D) The base is stronger than the acid.



3. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The acid is stronger than the base.
 (B) The base is stronger than the acid.
 (C) The acid is stronger than the base.
 (D) The base is stronger than the acid.



4. Which of the following is not a property of a strong acid? (correct)



5. A solution of a weak acid and its conjugate base is called a buffer solution. Which of the following is not a property of a buffer solution? (correct)
- (A) The pH is constant.
 (B) The pH is constant.
 (C) The pH is constant.
 (D) The pH is constant.



6. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The acid is stronger than the base.
 (B) The base is stronger than the acid.
 (C) The acid is stronger than the base.
 (D) The base is stronger than the acid.



Multiple Choice Stoichiometry Test

Guillaume Favre

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