

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. salt bridge (anion/cation flow) \_\_\_\_\_
- f. displacement (anode/cathode) \_\_\_\_\_
- g. silver-silver chloride electrode \_\_\_\_\_
- h. Zn \_\_\_\_\_
- i. buffer capacity and pH:  $\text{NH}_4^+/\text{NH}_3$  buffer in alkaline solution, explain the buffer components and how maintaining the original pH is done!

\_\_\_\_\_

# Voltaic Cells Study Guide

**S Ben Porath**



## **Voltaic Cells Study Guide:**

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