

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. Anode: _____
- b. Cathode: _____
- c. Anion: _____
- d. Cation: _____
- e. Half-reaction (oxidation) at anode: _____
- f. Half-reaction (reduction) at cathode: _____
- g. Net redox reaction: _____
- h. Salt: _____
- i. Write overall cell equation with phases. $\text{Zn(s)} + \text{Cu}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Cu(s)}$ (10 marks)

Voltaic Cells Study Guide

Peipei Pang



Voltaic Cells Study Guide:

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