

Name: \_\_\_\_\_

## Research Paper Rubric

	4	3	2	1
<b>Content and Focus</b>	<ul style="list-style-type: none"> <li>• Exceptionally clear, focused, interesting thesis.</li> <li>• Strong, rich supporting details and examples that prove thesis.</li> <li>• A meaningful conclusion explaining the importance of the research and how it can be used.</li> </ul>	<ul style="list-style-type: none"> <li>• Clear thesis which maintains a consistent focus from beginning to end.</li> <li>• Specific supporting details are present.</li> <li>• A clear conclusion as to why the research is important.</li> </ul>	<ul style="list-style-type: none"> <li>• Contains thesis but with inconsistent focus.</li> <li>• Generalized supporting details that prove thesis.</li> <li>• Conclusion tends to summarize research.</li> </ul>	<ul style="list-style-type: none"> <li>• Thesis statement lacks clarity and focus.</li> <li>• Inadequate or missing supporting details.</li> <li>• Missing a summarizing conclusion.</li> </ul>
<b>Organization</b>	<ul style="list-style-type: none"> <li>• Strong introduction and conclusion.</li> <li>• Consistent and coherent logical progression.</li> <li>• Uses clear and skillful transitions.</li> </ul>	<ul style="list-style-type: none"> <li>• Clear introduction and conclusion.</li> <li>• Illustrates some consistency and shows some logical progression.</li> <li>• Uses clear transitions.</li> </ul>	<ul style="list-style-type: none"> <li>• Introduction and conclusion is present but not clear.</li> <li>• Show some attempt of consistency and order. Paper shows attempt of transitions between paragraphs.</li> </ul>	<ul style="list-style-type: none"> <li>• Unable to clearly identify introduction and conclusion.</li> <li>• Lack of consistency and order.</li> <li>• Paper shows little or no attempt of transitions between paragraphs.</li> </ul>
<b>Style</b>	<ul style="list-style-type: none"> <li>• Written in formal language (avoids slang, completely).</li> <li>• Elaborate and colorful language.</li> <li>• Consistently strong and varied sentence structure.</li> <li>• Direct quotes support student's ideas.</li> <li>• Paper written in student's own words.</li> </ul>	<ul style="list-style-type: none"> <li>• Majority of paper written in formal language.</li> <li>• Language appropriate to topic.</li> <li>• Words convey intended message.</li> <li>• Direct quotes support student's ideas.</li> <li>• Majority of paper written in student's own words.</li> </ul>	<ul style="list-style-type: none"> <li>• Some use of formal language recognized; informal language is dominant.</li> <li>• Most language is appropriate to topic. Able to get vague idea of message.</li> <li>• Some parts of paper written in student's own words.</li> </ul>	<ul style="list-style-type: none"> <li>• Paper frequently uses informal language.</li> <li>• Language is not appropriate to topic.</li> <li>• Message is unclear.</li> <li>• Majority of paper is plagiarized.</li> </ul>
<b>Sources/Format</b>	<p>Follows MLA guidelines:</p> <ul style="list-style-type: none"> <li>• Uses 5 or more cited sources.</li> <li>• Sources meet the guidelines for types of sources.</li> <li>• All parenthetical documentation is MLA correct (author's name, pg. #).</li> <li>• Works Cited page is MLA correct.</li> <li>• All researched info is documented.</li> </ul>	<p>Follows MLA Guidelines with few exceptions:</p> <ul style="list-style-type: none"> <li>• 3-4 cited sources used.</li> <li>• Sources meet the guidelines for types of sources.</li> <li>• Few errors noted in parenthetical documentation.</li> <li>• Majority of Works Cited page is MLA correct.</li> <li>• Most research info is Documented.</li> </ul>	<ul style="list-style-type: none"> <li>• Inconsistent use of MLA style guidelines. Less than 3 cited sources used.</li> <li>• Majority of parenthetical documentation done incorrectly.</li> <li>• Random MLA documentation.</li> <li>• Rarely documents sources.</li> </ul>	<ul style="list-style-type: none"> <li>• Fails to follow MLA style Guidelines.</li> <li>• Less than 3 cited sources used little or no parenthetical documentation.</li> <li>• Works Cited page is not understandable.</li> </ul>
<b>Conventions</b>	<p>Superior editing. Rarely makes errors in the following areas:</p> <ul style="list-style-type: none"> <li>• Spelling and mechanics.</li> <li>• Correct usage and grammar.</li> </ul>	<p>Careful editing. Makes few errors in the following areas:</p> <ul style="list-style-type: none"> <li>• Spelling and mechanics</li> <li>• Correct usage and grammar</li> </ul>	<p>Some evidence of editing:</p> <ul style="list-style-type: none"> <li>• Extensive spelling and grammatical errors.</li> </ul>	<p>Poor editing:</p> <ul style="list-style-type: none"> <li>• Spelling and grammatical errors make it difficult to read paper.</li> </ul>

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**JG Myers**



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web  $m \times c \times \Delta t$  equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution  
and  $\Delta t$  is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of

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web the heat of reaction  $\Delta H_{\text{soln}}$  is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as  $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$   $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol}^{-1}$  rxn write equations to similarly represent the dissolving process for

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web hand warmers are usually placed inside the gloves to keep your hands warm the heat keeping your hands warm is formed as a result of an exothermic reaction which occurs in a small packet hand warmers have 5 main ingredients in them salt carbon polypropylene vermiculite water and iron

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web when the hand warmer pack is activated the solid dissolves in water and produces a large temperature change in the form of heat in this laboratory experiment the heat of solution will be calculated from the data for six salts in order to

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution  $\Delta H_{\text{soln}}$  or the amount of energy released or absorbed in kJ mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web feb 2 2017 this is a video outlining a flinn lab on designing a hand warmer it was created by alex brinley charis conwell and siena joy for our ap chemistry class

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web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

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