

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

Answer: B

16. According to kinetic theory of gases, the absolute zero temperature is attained when

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

Answer: C

17. Kinetic theory of gases assumes that the collisions between the molecules are

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

Answer: A

18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to

- A. E/3
- B. E/2
- C. 2E/3

Answer: A

19. The

20. Kinetic energy of the molecules in terms of absolute temperature T is proportional to

- A. T
- B. T²
- C. 1/T
- D. 1/T²
- E. T^{3/2}

Answer: A

21. The average velocity of gas molecules is

- A. directly proportional to \sqrt{T}
- B. directly proportional to T
- C. directly proportional to T^2
- D. directly proportional to T^3
- E. directly proportional to $T^{3/2}$

Answer: A

22. The average velocity of gas molecules is

- A. directly proportional to \sqrt{T}
- B. directly proportional to T
- C. directly proportional to T^2

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Robert H. Kim



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