

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance  
 (B) volume, molarity, pH, amount of substance  
 (C) volume, molarity, pH, concentration of ions  
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A)  $n = 1000$ ,  $m = 1000$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$   
 (C)  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

2. An acid-base reaction involves hydrogen ions. Suppose instead you consider hydroxide ions. If 1.0 g of acid were neutralized by 1.0 g of base, how many grams of acid would be neutralized by 1.0 g of base? (correct)

(A)  $n = 1000000$ ,  $m = 1000000$

(B)  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$

3. An acid-base reaction involves  $\text{H}^+$  ions.  $\text{H}^+$  ions are the smallest cations and are present in solutions of acids. (correct)

(A)  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$

4. Which of the following is not a property of the following equation? (correct)



(A)  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$

5. A solution of acid and base are mixed together and the pH is measured. If 1.0 g of acid and 1.0 g of base are mixed together, what mass of acid was neutralized? (correct)



(A)  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$

6. In a full equilibrium, acetic acid reacts with  $\text{H}_2\text{O}$  to form acetate and hydronium.  $\text{CH}_3\text{COOH} + \text{H}_2\text{O} \rightleftharpoons \text{CH}_3\text{COO}^- + \text{H}_3\text{O}^+$ . After the reaction was completed, the resulting solution was 1.0 g. What is the mass of acetate? (correct)



# Multiple Choice Stoichiometry Test

**Sasha Alcott**

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## **Table of Contents Multiple Choice Stoichiometry Test**

1. Understanding the eBook Multiple Choice Stoichiometry Test
  - The Rise of Digital Reading Multiple Choice Stoichiometry Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Stoichiometry Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Multiple Choice Stoichiometry Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Stoichiometry Test
  - Personalized Recommendations
  - Multiple Choice Stoichiometry Test User Reviews and Ratings
  - Multiple Choice Stoichiometry Test and Bestseller Lists
5. Accessing Multiple Choice Stoichiometry Test Free and Paid eBooks
  - Multiple Choice Stoichiometry Test Public Domain eBooks
  - Multiple Choice Stoichiometry Test eBook Subscription Services
  - Multiple Choice Stoichiometry Test Budget-Friendly Options

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  - ePub, PDF, MOBI, and More
  - Multiple Choice Stoichiometry Test Compatibility with Devices
  - Multiple Choice Stoichiometry Test Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Multiple Choice Stoichiometry Test
  - Highlighting and Note-Taking Multiple Choice Stoichiometry Test
  - Interactive Elements Multiple Choice Stoichiometry Test
8. Staying Engaged with Multiple Choice Stoichiometry Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Multiple Choice Stoichiometry Test
9. Balancing eBooks and Physical Books Multiple Choice Stoichiometry Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Multiple Choice Stoichiometry Test
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice Stoichiometry Test
  - Setting Reading Goals Multiple Choice Stoichiometry Test
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice Stoichiometry Test
  - Fact-Checking eBook Content of Multiple Choice Stoichiometry Test
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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