

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

Answer: B

16. According to kinetic theory of gases, the absolute zero temperature is attained when

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

Answer: C

17. Kinetic theory of gases assumes that the collisions between the molecules are

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

Answer: A

18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to

- A. E/3
- B. E/2
- C. 2E/3

Answer: A

19. The

Answer: A

20. The kinetic energy of the molecules in terms of absolute temperature T is proportional to

- A. T
- B. T²
- C. 1/T
- D. 1/T²
- E. T^{3/2}

Answer: A

21. The temperature of a gas is directly proportional to

- A. average kinetic energy
- B. average potential energy
- C. average total energy
- D. average momentum
- E. average mass of gas molecules

Answer: A

22. The absolute zero temperature is defined as

- A. the temperature at which the volume of a gas becomes zero
- B. the temperature at which the pressure of a gas becomes zero
- C. the temperature at which the kinetic energy of the molecules becomes zero

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**Dr. J. P. Goel, ,D. C. Upadhyay, ,Er.
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