

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. salt bridge (indicate flow) \_\_\_\_\_
- f. displacement (write balanced) \_\_\_\_\_
- g. half-cell reduction (write) \_\_\_\_\_
- h. Zn \_\_\_\_\_
- i. Write overall cell process:  $\text{Zn} + \text{Cu}^{2+} \rightarrow \text{Zn}^{2+} + \text{Cu}$  (do not forget to balance electrons, represent by either displacement or half-cell reduction/oxidation) \_\_\_\_\_

\_\_\_\_\_

# Voltaic Cells Study Guide

**M Carnoy**



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