

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a) anode _____
- b) cathode _____
- c) anion _____
- d) cation _____
- e) direction of electron flow _____
- f) direction of ion movement _____
- g) half-cell reduction equation _____
- h) cell _____
- i) Write overall cell equation with phases. $\text{cell} = \text{anode} | \text{anode solution} || \text{cathode solution} | \text{cathode}$ (10 marks)

Voltaic Cells Study Guide

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