

17.1

HYDROGEN IONS AND ACIDITY

Section Overview

Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and an appropriate equilibrium.
2. Calculate hydroxide and hydronium ion concentrations from K_b and K_a values and the concentration of a weak acid or base.
3. Calculate the pH of weak acid and base solutions.

Vocabulary

- | | | |
|-------------------|-----------------------------|-------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. weak acid/base |
| 4. hydronium ion | 5. K_a value | 6. K_b value |
| | 7. weak base | 8. pH |

Key Equations

- | | |
|--|--|
| 1. $K_a = \frac{[\text{H}_3\text{O}^+][\text{A}^-]}{[\text{HA}]}$ (Eq. 17.1) | 2. $K_b = \frac{[\text{B}][\text{OH}^-]}{[\text{BOH}]}$ (Eq. 17.2) |
| 3. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ (Eq. 17.3) | 4. $\text{pOH} = -\log[\text{OH}^-]$ (Eq. 17.4) |

Part A: Completion

Use the appropriate number of significant figures in the answers to the following questions. Show calculations for all problems that require calculations.

1. What products are **1** **2** formed by the autoionization of water? Write the autoionization reaction of water and the K_w value at 25°C.
2. Write the K_a expression for the acid dissociation of H_2SO_4 .
3. Write the K_b expression for the base dissociation of NH_3 .
4. Write the K_a expression for the acid dissociation of H_2SO_4 .
5. Write the K_b expression for the base dissociation of NH_3 .
6. Write the K_a expression for the acid dissociation of H_2SO_4 .
7. Write the K_b expression for the base dissociation of NH_3 .

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19 2 Hydrogen Ions And Acidity Worksheet Answers

Jessica J Manson



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