

Top Essential Oil & Their Uses

Oil	Uses	How To Use
 <p>Lavender</p>	<ul style="list-style-type: none"> • Help your anxiety • Help your insomnia • Treat your acne • Relieves itchy bug bites 	<ul style="list-style-type: none"> • Massage into your skin • Add to your bath • Sprinkle a few drops onto a cloth • Inhale its aroma • Add the oil to a diffuser
 <p>Peppermint</p>	<ul style="list-style-type: none"> • Relieve your sore muscles • Treat your nausea • Treat your headaches • Freshen your breath 	<ul style="list-style-type: none"> • Fill a bowl with a hot water oil mix and inhale vapor • Massage into your skin • Add 3 drops under your tongue for bad breath
 <p>Lemon</p>	<ul style="list-style-type: none"> • Use as a deodorizer • To stay focused • Help your skin heal • As a cleaner • Help your anxiety 	<ul style="list-style-type: none"> • Add to cleansers • Massage into your skin • Add the oil to a diffuser • Mix with water and baking soda to make a degreaser
 <p>Frankincense</p>	<ul style="list-style-type: none"> • Help gut your health • Ward off gum disease • Improve your skin 	<ul style="list-style-type: none"> • Disinfects • Keep your hair shiny • Balance your skin
 <p>Tea Tree</p>	<ul style="list-style-type: none"> • Treat you acne • Odor eliminator • Treat your dandruff 	<ul style="list-style-type: none"> • Soothe your skin irritation • Fights fungus • Use as mouthwash
 <p>Orange</p>	<ul style="list-style-type: none"> • Improve your mood • Boost your indigestion • Improve your cognitive function • Give your skin a glow 	<ul style="list-style-type: none"> • Use as a deodorizer • Treat your acne
 <p>Geranium</p>	<ul style="list-style-type: none"> • Treat your breakouts & skin irritation • Ease your nasal symptoms 	<ul style="list-style-type: none"> • Reduce your allergic reactions • Fight your bacterial infections • Treat your stress
 <p>Bergamot</p>	<ul style="list-style-type: none"> • Use for soothing • Aromatherapy • Spot treatment for your acne • Treat against painful cysts 	<ul style="list-style-type: none"> • To scent your hair • To soften and tame your curls
 <p>Grapefruit</p>	<ul style="list-style-type: none"> • Aromatherapy • Help balance your mood • Calming • Lower your blood pressure • Keeps you from getting acne 	<ul style="list-style-type: none"> • Support your healthy metabolism • Help with your weight loss • Fight hangovers
 <p>Eucalyptus</p>	<ul style="list-style-type: none"> • Help your cough • Clear Chest Congestion • Keep the bugs away • Fight your skin inflammation 	<ul style="list-style-type: none"> • Help your respiratory conditions • Soothe your cold sores • Freshen your breath
 <p>Chamomile</p>	<ul style="list-style-type: none"> • Helps with digestion • Help balance your mood • Ease your skin conditions 	<ul style="list-style-type: none"> • Heal your wounds • Help your sleep • Use for anti-inflammation
 <p>Basil</p>	<ul style="list-style-type: none"> • Ease minor skin irritations • Aromatherapy • Calming • Help balance your mood • Insect repellent 	<ul style="list-style-type: none"> • Brighten your complexion • Use to soften and strengthen your hair • Eliminate odor • Soothe your headaches
 <p>Ylang Ylang</p>	<ul style="list-style-type: none"> • Boost your mood • Lower your blood pressure • Repel flying insects • Promote new hair growth 	<ul style="list-style-type: none"> • Treat your stomach distress • Treat your headache • Reducing candida albicans • Clear your skin
 <p>Rosemary</p>	<ul style="list-style-type: none"> • Stimulates your hair growth • Reduce your scalp itching • Repels bugs • Eases your stress 	<ul style="list-style-type: none"> • Improve your mood • Help your digestive & liver health • Help reduce your joint inflammation
 <p>Melissa/Lemon balm</p>	<ul style="list-style-type: none"> • Relieve your stress • Balance your mood • Improve your cognitive function • Help insomnia 	<ul style="list-style-type: none"> • Treat cold sores • Relieve your indigestion • Help your nausea • Help ease your headache

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web the ions which do not participate in chemical reactions and present the same on both sides of the reactions are called
 spectator ions in the net chemical reaction the spectator ions are cancelled from both sides of the equation in other words
 spectate means watch the other ions react in an aqueous solution

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web answer $\text{Na}^+ \text{aq}$ and $\text{SO}_4^{2-} \text{aq}$ identify the chemical equation the spectator ions $\text{KOH} \text{aq}$ HNO_3 $\text{KNO}_3 \text{aq}$ H_2O $\text{K}^+ \text{aq}$ $\text{OH}^- \text{aq}$ $\text{H}^+ \text{aq}$
 $\text{NO}_3^- \text{aq}$ $\text{K}^+ \text{aq}$ $\text{NO}_3^- \text{aq}$ H_2O l answer $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ identify the chemical equation the spectator ions $\text{Pb}^{2+} \text{aq}$ $\text{PbI}_2 \text{s}$
 $2\text{KNO}_3 \text{aq}$

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions
 gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq}$
 $\rightarrow 2\text{Cr}^{3+} \text{aq} + 7\text{H}_2\text{O}$

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is $2 \text{Na}^+ \text{aq} + 2 \text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} + 2 \text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{ s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{ l} + \text{CO}_2 \text{ g}$ the spectator ions are Cl^- and Na^+
net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{ l} + \text{CO}_2 \text{ g} + \text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{H}_2\text{O} \text{ l} + \text{CO}_2 \text{ g}$ here is a video on spectator ions
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web the six $\text{NO}_3^- \text{aq}$ ions and the six $\text{Na}^+ \text{aq}$ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3\text{PO}_4 \text{ s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{ l}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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web jul 26 2017 answer the spectator ions are explanation in the net ionic equations we are not include the spectator ions in the equations spectator ions the ions present on reactant and product side 16 used to grind chemicals to powder tmraor nda stlepe star 4 4 5 heart 181 verified verified answer food

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web the new format requires students to write balanced chemical equations showing only the reacting substances excluding spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction the questions are intended to test the students understanding of the meaning of the reactions

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be 2cl aq cu 2 aq cucl 2 s 2cl aq cu 2 aq cucl 2 s these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

