

## REGLA DEL OCTETO



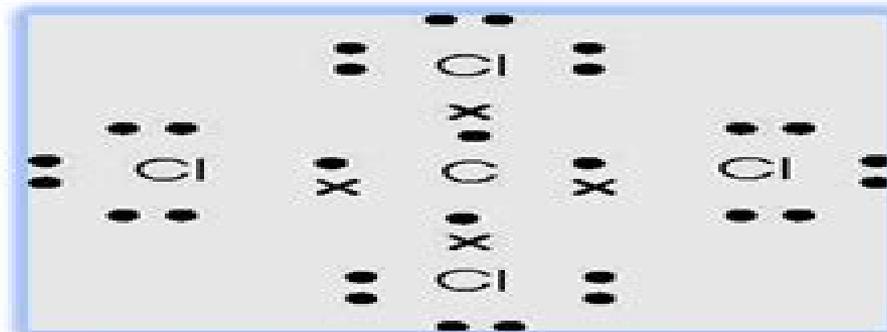
En la unión de estos átomos, el sodio (Na) transfiere su único electrón de valencia al cloro para que pueda completar su octeto



El cloro y azufre comparten sus electrones para completar su octeto (8 electrones de valencia)

## REPRESENTACIÓN DE ENLACES CON LA ESTRUCTURA DE LEWIS

Los electrones de los orbitales externos se indican por medio de puntos o cruces alrededor del elemento.



## Enlaces Text

**M Tight**



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