

The curve in Figure 16.11 of the text shows how the vibrational heat capacity depends on temperature. Note that even when the temperature is only slightly above the vibrational temperature, the heat capacity is close to its equipartition value.

The total heat capacity of a molecular substance is the sum of each contribution (Figure 16.12 of the text). When equipartition is valid (when the temperature is well above the characteristic temperature of the mode $T \gg \theta_M$) we can estimate the heat capacity by counting the numbers of modes that are active. In gases, all three translational modes are always active and contribute $\frac{3}{2}R$ to the molar heat capacity. If we denote the number of active rotational modes by ν_R^\ddagger (so for most molecules at normal temperatures $\nu_R^\ddagger = 2$ for linear molecules, and 3 for non-linear molecules), then

$$C_{V,m} = \frac{3}{2}R + \frac{\nu_R^\ddagger}{2}R + \dots$$

The equipartition value $C_{V,m} = \frac{3}{2}R + \frac{\nu_R^\ddagger}{2}R$ is reached when the temperature is high enough so that $T \gg \theta_M$ for all the modes. However, if the temperature is lower, some of the modes are not fully excited and the heat capacity is lower than the equipartition value. The curve in Figure 16.11 shows how the vibrational heat capacity depends on temperature. Note that even when the temperature is only slightly above the vibrational temperature, the heat capacity is close to its equipartition value.

Solutions to exercises

The vibrational heat capacity is given by $C_{V,m} = \frac{R}{\theta_M} \sum_{j=1}^{\infty} j e^{-j\theta_M/T}$. When $\theta_M \ll T$, the exponential can be expanded as $e^{-j\theta_M/T} \approx 1 - j\theta_M/T + \frac{1}{2}(j\theta_M/T)^2 - \dots$

$$C_{V,m} \approx \frac{R}{\theta_M} \sum_{j=1}^{\infty} j \left(1 - j\frac{\theta_M}{T} + \frac{1}{2}j^2\left(\frac{\theta_M}{T}\right)^2 - \dots \right)$$

and the sum of the series can be evaluated as follows:

$$\sum_{j=1}^{\infty} j = \frac{1}{\theta_M} \sum_{j=1}^{\infty} j \theta_M = \frac{1}{\theta_M} \theta_M \sum_{j=1}^{\infty} 1 = \frac{1}{\theta_M} \theta_M \frac{1}{\theta_M} = \frac{1}{\theta_M}$$

and similarly

$$\sum_{j=1}^{\infty} j^2 = \frac{1}{\theta_M^2} \sum_{j=1}^{\infty} j^2 \theta_M^2 = \frac{1}{\theta_M^2} \theta_M^2 \sum_{j=1}^{\infty} j^2 = \frac{1}{\theta_M^2} \theta_M^2 \frac{1}{\theta_M} = \frac{1}{\theta_M}$$

$$C_{V,m} \approx \frac{3}{2}R + \frac{\nu_R^\ddagger}{2}R + \frac{1}{2}R \left(\frac{\theta_M}{T} \right)^2 - \dots$$

which is the result in the text.

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Mike Jess



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web wilflex ink management software ims has been an essential component in textile ink rooms since the late 1980s when wilflex launched its first pc based system the new ims 3 0 system was developed to better support all wilflex color system solutions which now include automated dispensing smart scale integration and best in class systems

inks wilflex epic standard colors avient - Jan 08 2023

web wilflex epic pfx standard colors have excellent press performance high speed wet on wet printing a smooth satin finish a

proven technology platform and are available for a wide range of colors simple to use and easy to print wilflex epic inks are based on well known and time tested formulations of inks

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web wilflex ink formulations for screen printing produce simulations of pantone g colors printed results may vary based on production methods such as ink film thickness opacity pigment selection and substrate pantone is pantone inc s check standard for color portions c pantone inc 1963 1991 pantonc is a trademark of pantone inc

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web the third column is the closest matching ink found on the wilflex color chart these are the most commonly used inks the last column is the closest matching pantone color this is used in case you need to find the closest possible pms for a vinyl color

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web jun 9 2021 by keeping the colors separate you re making your life easier in case you add too much of a color you ll be able to remove a bit of the ink without scooping out the other colors let s talk about color matching when you have the color mixed it s time to confirm if it s accurate

wilflex avient specialty inks - Nov 06 2022

web the wilflex brand of inks emerged in the early 1960s as screen printing on t shirts gained popularity for self expression allowing the wilflex brand to gain global recognition wilflex continued to make history twenty years later when it provided design freedom allowing hundreds of color options with the introduction of the first plastisol