

Large Print Word Search Puzzle

This is large print general knowledge non-themed word search puzzle of easy level. Find in the grid and cross out all the words from the word list. Ignore spaces, dashes and diacritics, if any. Words go horizontally, vertically, not diagonally, may intersect, do not bend, overlap or go backward. There are no unused letters in the grid.

APPLE
 AWARD
 BEACH
 BOTTLE
 CAMEL
 CELLAR
 ELEGANT
 GUARANTEE
 HELMET
 ISLAND
 JACKET
 JELLYFISH
 LARGE
 LEISURE
 MOLECULE
 PENCIL
 POTATO
 REASON
 REINDEER
 RIDDLE

SADDLE
 SALMON
 SLICE
 STONE
 TEAM
 TELL
 THREE
 TURTLE
 UNIQUE
 VOYAGE

J	A	C	K	E	T	S	A	D	D	L	E
E	L	E	G	A	N	T	A	W	A	R	D
L	P	O	T	A	T	O	C	A	M	E	L
L	M	H	T	B	H	N	S	I	V	I	G
Y	O	E	U	O	R	E	A	S	O	N	U
F	L	L	R	T	E	L	L	L	Y	D	A
I	E	M	T	T	E	A	M	A	A	E	R
S	C	E	L	L	A	R	O	N	G	E	A
H	U	T	E	E	P	G	N	D	E	R	N
S	L	I	C	E	P	E	N	C	I	L	T
B	E	A	C	H	L	E	I	S	U	R	E
R	I	D	D	L	E	U	N	I	Q	U	E

Answer:

J	A	C	K	E	T	S	A	D	D	L	E
E	L	E	G	A	N	T	A	W	A	R	D
L	P	O	T	A	T	O	C	A	M	E	L
L	M	H	T	B	H	N	S	I	V	I	G
Y	O	E	U	O	R	E	A	S	O	N	U
F	L	L	R	T	E	L	L	L	Y	D	A
I	E	M	T	T	E	A	M	A	A	E	R
S	C	E	L	L	A	R	O	N	G	E	A
H	U	T	E	E	P	G	N	D	E	R	N
S	L	I	C	E	P	E	N	C	I	L	T
B	E	A	C	H	L	E	I	S	U	R	E
R	I	D	D	L	E	U	N	I	Q	U	E

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WJ Hussar



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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and

explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

nitration of bromobenzene 6 nitration of bromobenzene 10 19 - Oct 01 2023

over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 M H_2SO_4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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lab report 15 nitration of bromobenzene brynn abbott amp katie muerer bea0006 uah ch 336 02 september 21 2022 abstract in this experiment the electrophilic aromatic substitution mechanism was used to obtain 4-nitrobromobenzene and 2-

nitrobromobenzene recrystallization was used to purify 4-nitrobromobenzene

nitration of bromobenzene the purpose of this lab was to - Jul 30 2023

the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o-bromonitrobenzene

and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

lab report nitration of bromobenzene ipl org - Feb 22 2023

lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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web aug 19 2023 we are here going to cover up the aiken county public schools schools calendar of south carolina state schedule to help the follow scholars of the city the majority of the aiken county public schools in south

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web jun 26 2023 this calendar outlines the academic year including start and end dates holidays breaks and other events it helps everyone plan their schedules accordingly and stay on top of important deadlines one key feature of the aiken county school calendar is its emphasis on professional development days for teachers

2021 22 aiken county public school district calendar acpsd - Jul 17 2023

web memorial day holiday half day for students june 1 2 2 3 2 3 half day for students end of 2nd semester last day of school teacher work day graduation dates 7 10 24 q4 final report cards 4 10 s schools and offices closed

aiken county school board to discuss modified calendar for 2022 - Aug 06 2022

web jan 10 2022 the draft of the 2022 23 modified calendar for aiken county the aiken county school board meets at 6 p m tuesday at the district office 1000 brookhaven drive aiken

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web aug 25 2023 this page contains the major holiday dates from the 2023 and 2024 school calendar for aiken county public schools in south carolina please check back regularly for any amendments that may occur or consult the aiken county public schools website for their 2023 2024 approved calendar

aiken county school board will consider 2022 23 calendar at - Jun 04 2022

web feb 7 2022 spring break would be april 3 7 2023 the last day of school would be thursday june 8 for students and friday june 9 2023 for teachers graduation would be june 7 8 and 9 the agenda notes

aiken county school leaders finalize calendar for 2023 24 year wrdw - Apr 14 2023

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web jun 21 2023 aiken county public schools calendar 2024 2025 june 21 2023 by mycollegepoints for schools to coordinate and communicate with all of the students and parents of various classes and in different schools the school calendar is essential

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web 30 31 holiday memorial day half day for students half day for students graduation half day for students graduation last day of school end of quarter 4 report cards

edgefield county schools unveil 2024 2025 modified academic calendar - Oct 28 2021

web nov 17 2023 in 2023 aiken county public school district also adopted the modified calendar with july start dates for more details on the 2024 2025 ecsd calendar visit the school district website

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web feb 9 2022 the aiken county board of education approved the calendar for the 2022 23 school year at its meeting tuesday night the approved calendar has teachers returning on monday aug 8 and students

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