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A Novel by ANTHONY STOREY



Graceless Go I

Arturo Cuomo



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web a reaction and its experimental rate law are represented below $2 \text{ ICl}_g + \text{H}_2 \text{ g} \rightarrow 2 \text{ HCl}_g + 2 \text{ I}_g$ rate $k [\text{ICl}]^2 [\text{H}_2]$ a chemist proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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consider the following mechanism
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ fast
 $\text{O} + \text{O}_2 \rightarrow \text{O}_3$ slow
write the overall balanced chemical equation
identify any intermediates within the mechanism

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 $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
the reaction between Mg and HCl is represented by the equation above in a kinetics experiment
a 0.080 g sample of Mg ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M HCl at 25 °C

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reactions conducted in an industrial facility mix compounds together heating and stirring them for a

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