

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities (correct)?  
 (A) pressure, concentration (B) pressure, pH  
 (C) molar fraction (D) molar fraction, volume
2. Which of the following is CORRECT? (A combination of two dimensional quantities)

(A)  $\text{pH} = \text{pH}_1 + \text{pH}_2$

(B)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1$  (C)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_2$  (D)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$  (E)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$

3. An acid-base indicator has two forms: un-ionized and ionized. The un-ionized form is colorless and the ionized form is pink. The  $\text{pH}$  of a solution is 7.0. What color will the solution be?

(A)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1$  (B)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_2$

(C)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$  (D)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$  (E)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$

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5. Which of the following is CORRECT? (A combination of two dimensional quantities)

(A)  $\text{pH} = \text{pH}_1 + \text{pH}_2$  (B)  $\text{pH} = \text{pH}_1 + \text{pH}_2$

(C)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1$  (D)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_2$  (E)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$

6. A solution of an acid and its conjugate base is prepared. The  $\text{pH}$  of the solution is 7.0. What color will the solution be?

(A)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1$  (B)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_2$  (C)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$  (D)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$

(E)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$  (F)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$  (G)  $\text{pH} = \text{pH}_1 + \frac{1}{2} \log C_1 C_2$

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