

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are the characteristic properties for a weak acid? _____
- (A) pK_a and K_a (B) pK_a and pH (C) K_a and pH (D) pK_a and K_b
2. Which of the following is CORRECT? _____



3. In a 0.1 M solution, which of the following is the most abundant species? _____
- (A) H_2O (B) H_3O^+ (C) OH^- (D) H_2O_2



4. In a 0.1 M solution, which of the following is the most abundant species? _____
- (A) H_2O (B) H_3O^+ (C) OH^- (D) H_2O_2



5. Which of the following is the most abundant species in a 0.1 M solution? _____



6. A solution of 0.1 M acetic acid and 0.1 M sodium acetate is prepared. The pK_a of acetic acid is 4.75. The pH of the solution is _____.



7. In a 0.1 M solution, which of the following is the most abundant species? _____
- (A) H_2O (B) H_3O^+ (C) OH^- (D) H_2O_2



Multiple Choice Stoichiometry Test

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