

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities (correct)?
 (A) pressure, concentration (B) pressure, pH
 (C) molar fraction (D) molar fraction, volume
2. Which of the following is CORRECT? (A) concentration of ions increases as pH increases (B) concentration of ions increases as pH decreases



3. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. If the pH of the solution is 4.0, what is the concentration of acetic acid?



4. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. What is the concentration of acetic acid?



5. Which of the following is CORRECT? (A) concentration of ions increases as pH increases (B) concentration of ions increases as pH decreases



6. A solution of acetic acid and its conjugate base is prepared. The pH of the solution is 4.0. If the concentration of acetic acid is 0.1 M, what is the concentration of acetate ions?



7. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. What is the concentration of acetic acid?



Multiple Choice Stoichiometry Test

Christian G. Meyer



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