

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are the characteristic properties for a weak acid? \_\_\_\_\_
- (A) pKa and molar concentration (B) pKa and pH
- (C) molar concentration (D) molar concentration and pH

2. Which of the following is CORRECT? In a weak acid solution, the concentration of the undissociated acid is \_\_\_\_\_
- (A)  $10^{-4}$  mol/L (B)  $10^{-5}$  mol/L
- (C)  $10^{-6}$  mol/L (D)  $10^{-7}$  mol/L
- (E)  $10^{-8}$  mol/L

3. In a 0.1 M weak acid solution, the concentration of the undissociated acid is 0.09 M. The pH of the solution is \_\_\_\_\_
- (A) 4.0 (B) 5.0 (C) 6.0 (D) 7.0 (E) 8.0

4. In a 0.1 M weak acid solution, the concentration of the undissociated acid is 0.09 M. The pH of the solution is \_\_\_\_\_
- (A) 4.0 (B) 5.0 (C) 6.0 (D) 7.0 (E) 8.0

5. Which of the following is CORRECT? In a weak acid solution, the concentration of the undissociated acid is \_\_\_\_\_
- (A)  $10^{-4}$  mol/L (B)  $10^{-5}$  mol/L (C)  $10^{-6}$  mol/L (D)  $10^{-7}$  mol/L (E)  $10^{-8}$  mol/L

6. Which of the following is CORRECT? In a weak acid solution, the concentration of the undissociated acid is \_\_\_\_\_
- (A)  $10^{-4}$  mol/L (B)  $10^{-5}$  mol/L (C)  $10^{-6}$  mol/L (D)  $10^{-7}$  mol/L (E)  $10^{-8}$  mol/L

7. In a 0.1 M weak acid solution, the concentration of the undissociated acid is 0.09 M. The pH of the solution is \_\_\_\_\_
- (A) 4.0 (B) 5.0 (C) 6.0 (D) 7.0 (E) 8.0

8. In a 0.1 M weak acid solution, the concentration of the undissociated acid is 0.09 M. The pH of the solution is \_\_\_\_\_
- (A) 4.0 (B) 5.0 (C) 6.0 (D) 7.0 (E) 8.0



# Multiple Choice Stoichiometry Test

**Xiaolong Qi**



**Multiple Choice Stoichiometry Test:**

## **Multiple Choice Stoichiometry Test** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Multiple Choice Stoichiometry Test**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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