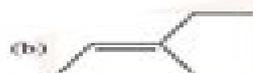


## GENERAL ORGANIC CHEMISTRY

### Tutorial – 7

#### Geometrical Isomerism : Cahn-Ingold-Prelog Priority Rules, E/Z Isomers, Counting Total Number of Isomers

1. Write (E) and (Z) configuration of the following :



2. Write structures of

(a) (Z)-1,3-pentadiene

(b) (E)-1,3-pentadiene

(c) (2E, 4E)-3-ethyl-2,4-hexadiene

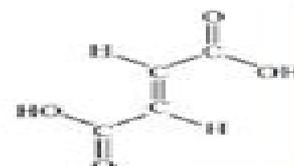
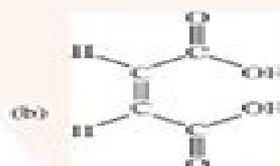
(d) (2Z, 4E)-3-ethyl-2,4-hexadiene

3. Compare dipole moments of

(I)



(II)



(iii) For each alkene, indicate the direction of the dipole moment. For each pair, determine which compound has the larger dipole moment.

(a) cis-1,2-difluoroethene or trans-1,2-difluoroethene

(b) cis-1,2-dibromoethene or trans-2,3-dibromo-2-butene

(c) cis-1,2-dibromo-1,2-dichloroethene or cis-1,2-dichloroethene

4. Draw all the different structures possible for hept-2,4-diene.

5. An organic compound with their molecular formula  $C_2H_2Cl_2$  exist in three isomeric forms A, B and C. Also A and C are polar, with A having greater dipole moment than C. B is non polar. Draw the structural formula of A, B and C and determine the isomeric relations between the pair, A, B and A, C and B, C.

6. Which of the following compounds will exhibit geometrical isomerism ?

(a) 1-Phenyl-2-butene

(b) 3-Phenyl-1-butene

(c) 2-Phenyl-1-butene

(d) 1, 1-Diphenyl-1-propene

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