

General Chemistry Practice Questions & Answers

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1. What is the atomic number of an element?

- The number of protons in the nucleus of an atom.

2. Define ionic bond.

- A bond formed by the transfer of electrons from a metal to a nonmetal.

3. How do you calculate molar mass?

- Sum of the atomic masses of all atoms in a molecule (g/mol).

4. Balance the equation: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

- $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

5. State the Ideal Gas Law equation.

- $PV = nRT$, where P = pressure, V = volume, n = moles, R = gas constant, T = temperature.

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