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A GRANT COUNTY THRILLER

A FAINT COLD FEAR

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Faint Cold Fear A Novel

M Tight



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alcohols ethanol 2 methylpropan 2 ol butan 2 ol but 2 en 1 ol $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ ii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iv $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ ch 3 6 aldehydes and ketones draw the structures for the following compounds ethanal butanone also known as butan 2 one but the 2 is unnecessary and is often left out 2 methylpropanal

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question 1 basic hydrocarbon nomenclature part a enter the number of carbons indicated by each of the following root names in the designated answer box part b the line formula for a branched alkane is shown below part c the line formula for another branched alkane is

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jan 21 2020 the point of attachment is number one and any other smaller groups are named as substituent groups on the larger group this numbering is independent of the numbering of the parent chain try to name the following compounds

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dec 16 2021 so the complete name is 5 bromo 7 chloro 6 hydroxy 2 2 5 trimethyl 7 octen 4 one 4 it is not difficult to find the parent structure for this compound which is a cyclic alcohol so the last name is cyclopropanol the naming of the substituent with the benzene ring is

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solutions 1 aromatic compound iodobenzene 2 alkane 3 methylhexane 3 ketone 4 alkane alkyl halide 3 chloro 4 ethyl 2 4

dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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1 carbon methane 2 carbons ethane 3 carbons propane 4 carbons butane 5 carbons pentane 6 carbons hexane 7 carbons heptane 8 carbons octane 9 carbons nonane 10 carbons decane while many of these names share a greek root with

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increases b decreases c remain the same d wider 3