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**Minjie Lin**



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web since the mole fraction of the o<sub>2</sub> is 0.050 it contributes 0.050 or if you will 5.0% of the total pressure thus total pressure is 0.210 / 0.050 = 4.2 atm problem 9 a sample of 1.43 g of helium and an unweighed quantity of o<sub>2</sub> are mixed in a flask at room temperature

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