

100 Acupressure Points For Disease Treatment in Thai Traditional Medicine

- 2 Fengkula, Sutiyekula, Right
- 3 Kalakharas, Fingers and Toes 20
- 4 Dwanas, Right Eye Root
- 5 Rucham, Right Ear Root
- 6 Sikkhina, Uvula
- 7 Kula-ya, Right Temple Pain
- 8 Paribhara, Shoulder Blades
- 9 Relieves Yawning and Belching
- 10 Relieves A Restless Mind
- 11 Relieves Drowsiness and Fatigue
- 12 Relieves Sodborg and Wind Hitting Chest
- 13 Relieves Drowsiness
- 14 Relieves Bitterness in The Mouth
- 15 Relieves Sweetness in The Mouth
- 16 Relieves Loss of Taste
- 17 Relieves Yaking While Sleeping
- 18 Relieves Swollen in The Right Ear
- 19 Kula-ya, Toad Hand and Foot
- 20 Relieves Stiff Tongue Symptoms
- 21 Relieves Stuffy Nose
- 22 Relieves Symptoms of Clogged Milk Glands
- 23 Relieves Abscesses in The Breast
- 24 Relieves Symptoms of Milk Not Latching
- 25 Relieves Leg Ache
- 26 Relieves Insomnia
- 27 Relieves Chronic Testicular Inflammation
- 28 Solve Sunstroke
- 29 Solve Chikhaun wind
- 30 Solve Drain Out The Fire Element
- 31 Solve Unable to Move
- 32 Rucham Moves Out
- 33 Relieves Shin Stiffness
- 34 Relieves Chronic Shin Inflammation
- 35 Relieves Leg Cramps and Spasms
- 36 Relieves Chronic Inflammation



- 37 Relieves Left Temple Pain
- 38 Relieves Nausea
- 39 Relieves Nasal Congestion
- 40 Relieves Junction Wind On The Stomach or Bowels
- 41 Relieves Softness in The Mouth, Muror in The Mouth
- 42 Relieves Long Lasting Yawning
- 43 Cures Drafts
- 44 Relieves Nasal Tightness
- 45 Cures Congested Milk
- 46 Relieves Heat Symptoms
- 47 Relieves Nasal Congestion
- 48 Relieves Cold in The Mucosa Area
- 49 Solve The Lack of Strength to Push
- 50 Relieves Inability to Defecate
- 51 Relieves Aches and Pains in The Thighs
- 52 Relieves Swollen Inguinal Glands
- 53 Relieves Heat in The Chest
- 54 Relieves Body Heat
- 55 Relieves Chest Tightness
- 56 Relieves Chest Collie Symptoms
- 57 Solve Stiff knee joints
- 58 Relieves Fire Cramps
- 59 Relieves Tingling
- 60 Solve Stiff knee joints
- 61 Relieves Shin Stiffness
- 62 Solve Cold Feet
- 63 Solve Leg Spasms
- 64 Relieves The Feeling of Hot Soles of The Feet
- 65 Solve Wind Circulation Problems



- 1 Relieve Fontanel Pain
- 66 Fengkula Moves Out
- 67 Relieves Frontal Pain
- 68 Solves Blurred Vision, Dizziness
- 69 Sikkhina Moves Out
- 70 Dwanas Moves Out
- 71 Relieves Chest Tightness
- 72 Relieves Chest Heat
- 73 Relieves Chest Collie
- 74 Relieves Dysentery Pain Wind
- 75 Relieves Cold Symptoms
- 76 Relieves Heat Symptoms
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- 99 Relieves Plantar Fasciitis
- 100 Relieves The Feeling of Heat On the Back of The Feet

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anchorage means a place a fixing or a fixture to which a lifeline or lanyard is connected for fall restraint or fall arrest 1 3 3

authorised manager means a person appointed by his employer or the principal as the competent

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web 5 this regulation shall apply only to work at height carried out on or after 1st may 2014 training for persons at work 6 it shall be the duty of the responsible person of any person who carries out or is to carry out any work at height to ensure that the person shall work at height in a workplace only after he has

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web jul 21 2022 select the safety equipment like lanyards and deceleration devices that will support you at the height you re working from consider your body height the fall distance sags in your harness and other equipment and the length of your lanyard

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web 2 3 1 lack of safe work procedures 43 of the cases studied did not have proper safe work procedures in place there are procedures required for carrying out any work safely such as planning for a specific job task training workers on a step by step work process of how a job can be done in as safe a manner as possible and emergency

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this video accompanies the chm 252 lab nitration of bromobenzene in this video students will get an introduction to the lab as

well as the set up for conducting the reaction students should

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho/para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho/para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno₃ h₂so₄ in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by nitration of bromobenzene

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nitrobromobenzene recrystallization was used to purify 4-nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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