



# Ferries In Scotland

**DP Hallahan**



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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the unknown slat burned orange but also had a slight blue color close to the stick

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web each element had a unique flame spectrum much like its own colored line fingerprint as with any suspect the fingerprint identified the element beyond question this method of using light to make identifications became known as spectroscopy principle of atomic absorption emission spectroscopy flame test lab ldc

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web oct 15 2022 the objectives of this lab are to perform light checks of metal cations in order to observe their typical colors apply graphics to determine the periodicity and energizer of the emissions 5 flame tests and atomic spectra experiment chemistry libretxts flame test formal lab report sc4 ldc 11 how can atomic

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following solutions lino3 cu no3 2 sr no3 2 ba no3 2 kno3 and nano3 since nitrates do not emit color you

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web task use a flame test to determine which ion salt produces the red color then using a spectroscope match the bright line spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material list causes red fireworks procedure

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web oct 15 2022 the objectives of this lab are to perform flame tests of metal cations in order to observe their characteristic colors perform calculations for determines the output and energy of the emitted

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web energy and matter answered worksheet the experiment conducted aimed to give students the opportunity to determine different metal ions in liquid and solid compounds by observing the color of the flames they give off when

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different metal cations the characteristic colors observed are due to emitted electromagnetic radiation from the excited metal cations in this lab how do the metal cations become excited

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web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

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web 25 october 2022 objective experiment 1 to show how flame test can be used to identify some metals ion 2 to identify some unknown metal ions using flame test introduction the arrangement of electrons around the nucleus of the atom is

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